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MAJOR CHEMICAL CONSTITUENTS OF *STROBILUS LUPULI*

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**Abstract:** *Strobilus Lupuli* consists of the dried strobiles or inflorescences of the female plants of *Humulus lupulus* L (Cannabaceae). We want to learn Major chemical constituents of *Strobilus Lupuli* and uses in traditional medicine.

**Key word:** *Strobilus Lupuli*, *Humulus lupulus* L, Cannabaceae.

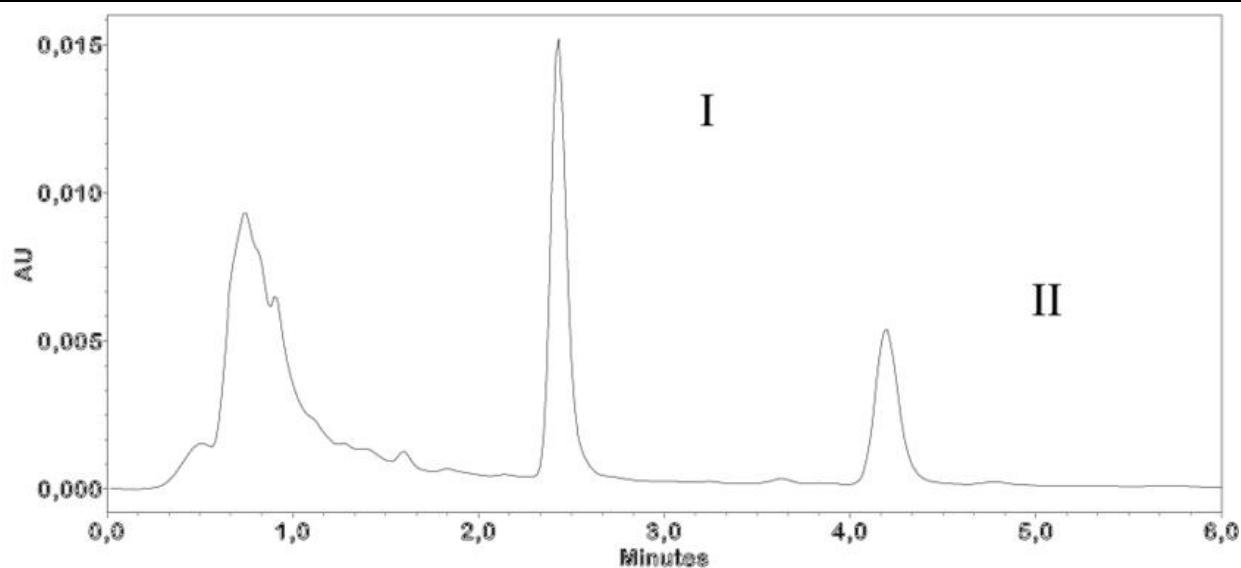
**Description.** A perennial, dioecious, twining herb, up to 6 m high. Aerial parts consist of several long, angular, rough-hairy, entwining stems bearing cordate, palmate, three-lobed, occasionally five- to seven-lobed, scabrous, dark green, stipulate leaves. Staminate flowers, with five bracts and five stamens, borne in axillary panicles. Pistillate flowers pale green, each consisting of an entire cup-like perianth and a unilocular ovary with a single ovule, and two long stigmas, borne on a leafy conical catkin. Fruits are ovate to ovate-cylindrical strobiles consisting of a flexuous rachis bearing yellowish-green to pale brown, ovate, membranous, scaly bracts, each enclosing a brown glandular achene.

**Plant material of interest: dried strobiles. General appearance**

Strobiles ovoid-cylindrical or cone-like, leafy, 3–4 cm long and up to 3 cm wide, consisting of a narrow, hairy, flexuous rachis and numerous imbricated, yellowish-green to dusky yellow, obliquely ovate, membranous bracts, the base of each with numerous orange to yellowish-orange, glandular trichomes, and frequently infolded on one side, enclosing a light brown subglobular glandular achene.

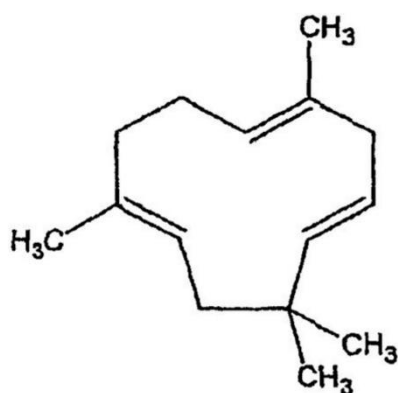
**Chemical assays**

High-performance liquid chromatography for bitter substances and Xanthohumol.

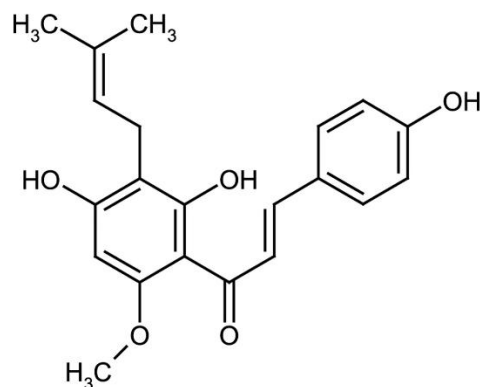


Representative HPLC chromatogram of a real extract sample: *Humulus lupulus*. Denotation of peaks: I-quercetin, II-kaempferol, isorhamnetin-detected only in traces. Column XBridge, mobile phase 0.1 % formic acid in water/methanol (14:11)

**Major chemical constituents** The major constituents are bitter substances (15–25%) in the resins. The resins are differentiated into hard (petroleum-ether insoluble) and soft resins. The lipophilic soft resins contain mainly  $\alpha$ -acids (e.g.  $\alpha$ -humulene (2,6,9-humulatriene) and related humulones) and  $\beta$ -acids (lupulones). The major chemical constituents of the soft resins are humulone and lupulone and their related derivatives, 2–10% and 2–6%, respectively. The hard resin contains a hydrophilic fraction,  $\delta$ -resin, and a lipophilic fraction,  $\gamma$ -resin. The essential oil (0.3–1.0%) contains mainly monoterpenes and sesquiterpenes such as  $\beta$ -caryophyllene, farnesene, humulene and  $\beta$ -myrcene. The essential oil also contains traces of 2-methylbut-3-ene-2-ol, which increases in amount to a maximum of 0.15% after storage of the strobiles for 2 years, owing to degradation of the humulones and lupulones. Other constituents include the chalcone xanthohumol, prenylflavonoids and other flavonoids (e.g. kaempferol, rutin) and tannins. Representative structures are presented below.



$\alpha$ -humulene



### xanthohumol

**Uses described in pharmacopoeias and well established documents** As a sedative for the treatment of nervous tension and insomnia. Treatment of dyspepsia and lack of appetite.

**Uses described in traditional medicine** Treatment of abdominal cramps, anaemia, bacterial infections, dermatitis, diarrhoea, dysmenorrhoea, leukorrhoea, migraine and oedema. As an analgesic, anthelmintic, antipyretic, carminative, depurative, digestant, diuretic, diaphoretic and tonic.

#### References:

1. Benkherouf, Ali Y.; Soini, Sanna L.; Stompor, Monika; Uusi-Oukari, Mikko (February 2019). "Positive allosteric modulation of native and recombinant GABAA receptors by hops prenylflavonoids". *European Journal of Pharmacology*. 852: 34–41.
2. European pharmacopoeia, 3rd ed. Suppl. 2001. Strasbourg, Council of Europe, 2001.
3. Farnsworth NR, ed. NAPRALERT database. Chicago, IL, University of Illinois at Chicago, 9 February 2001 production (an online database available directly through the University of Illinois at Chicago or through the Scientific and Technical Network (STN) of Chemical Abstracts Services).